

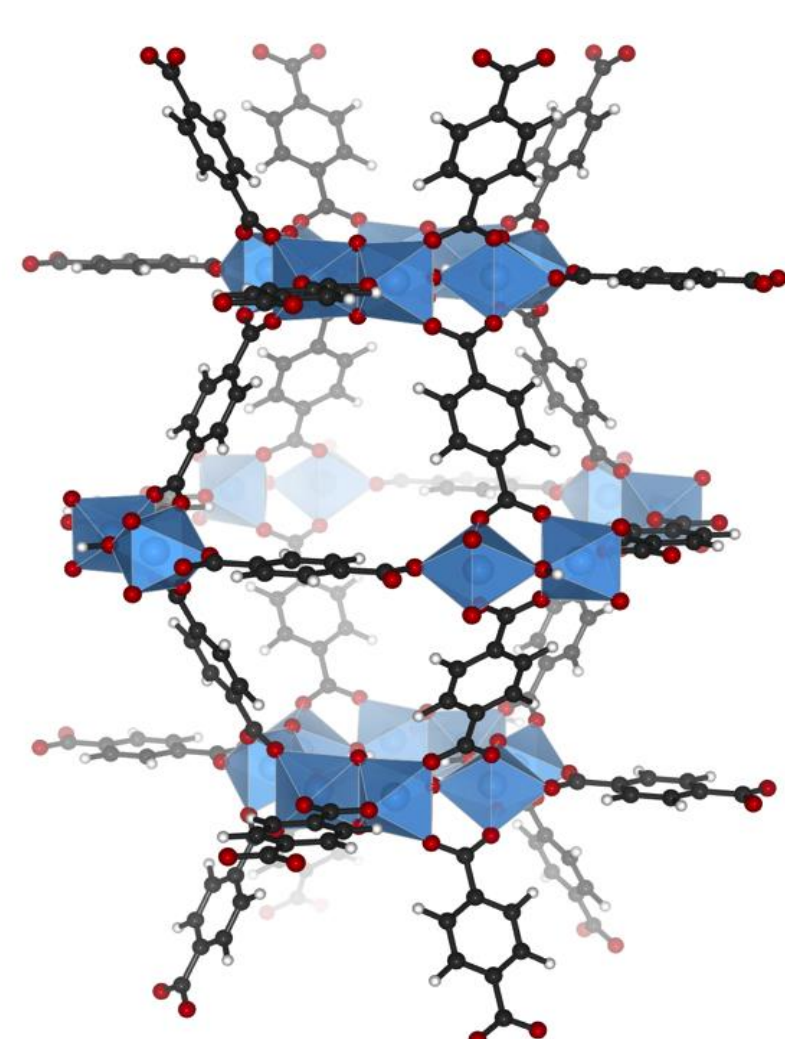
2D conductive MOF electronic property study

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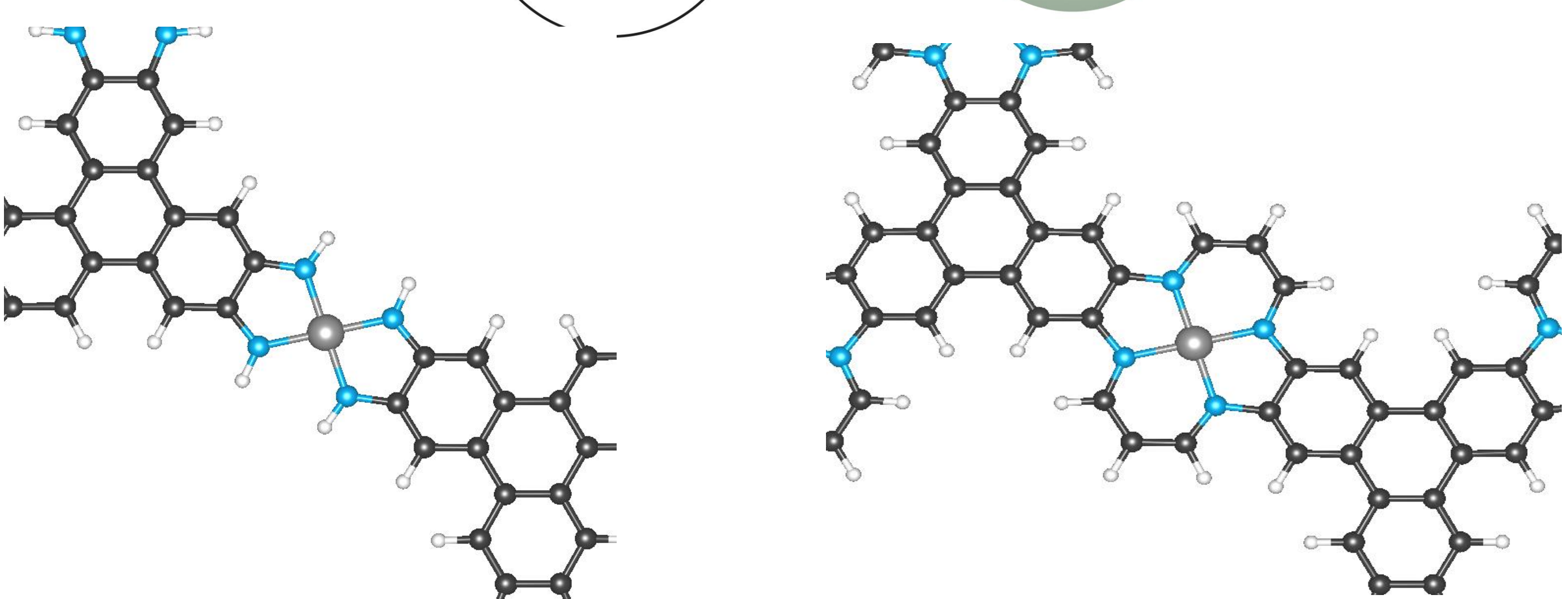
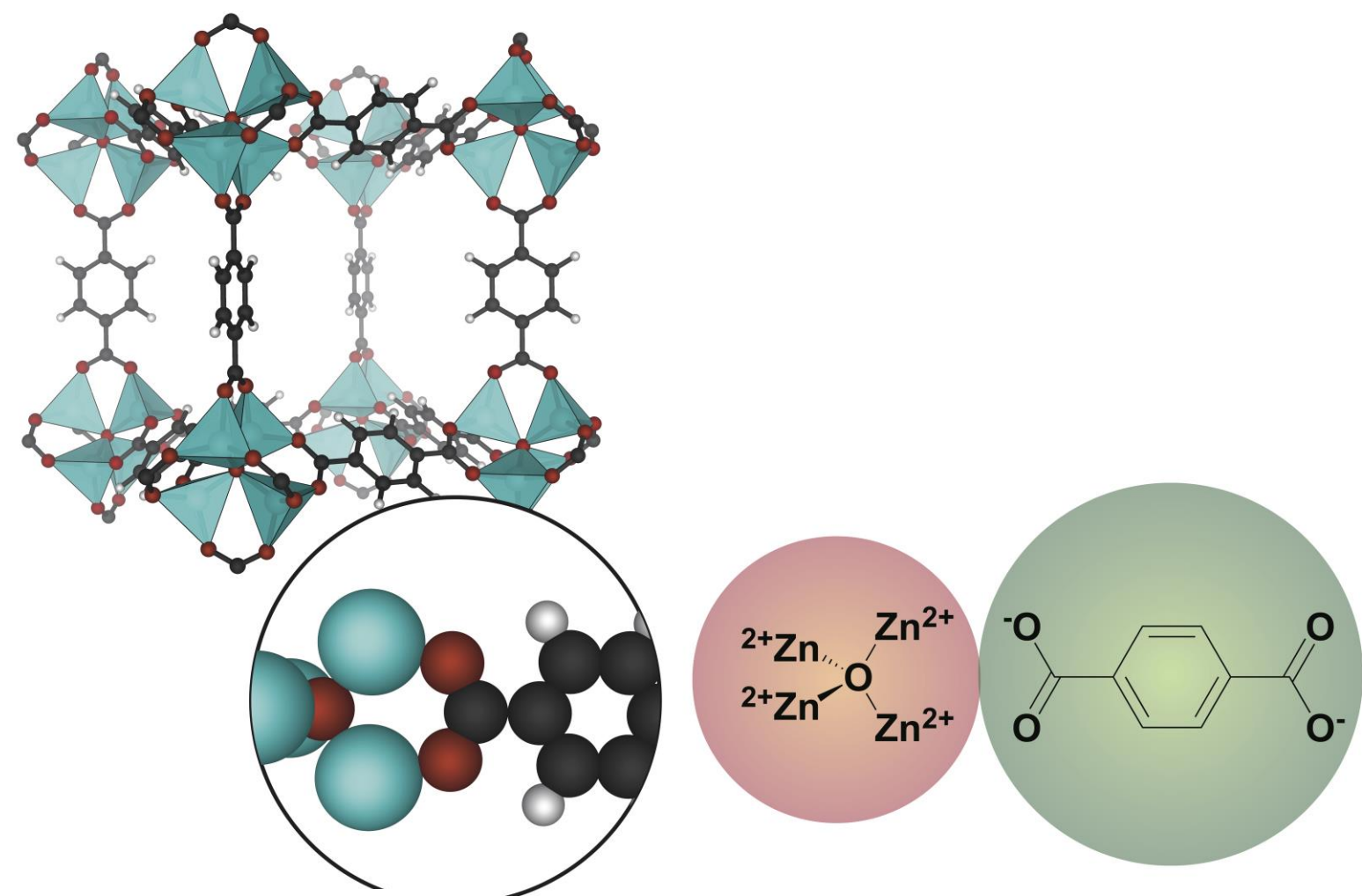
Metal Organic Framework

Gas storage
Electro-catalysis
Energy storage
Sensors
Electronic devices



Charge localization problem:

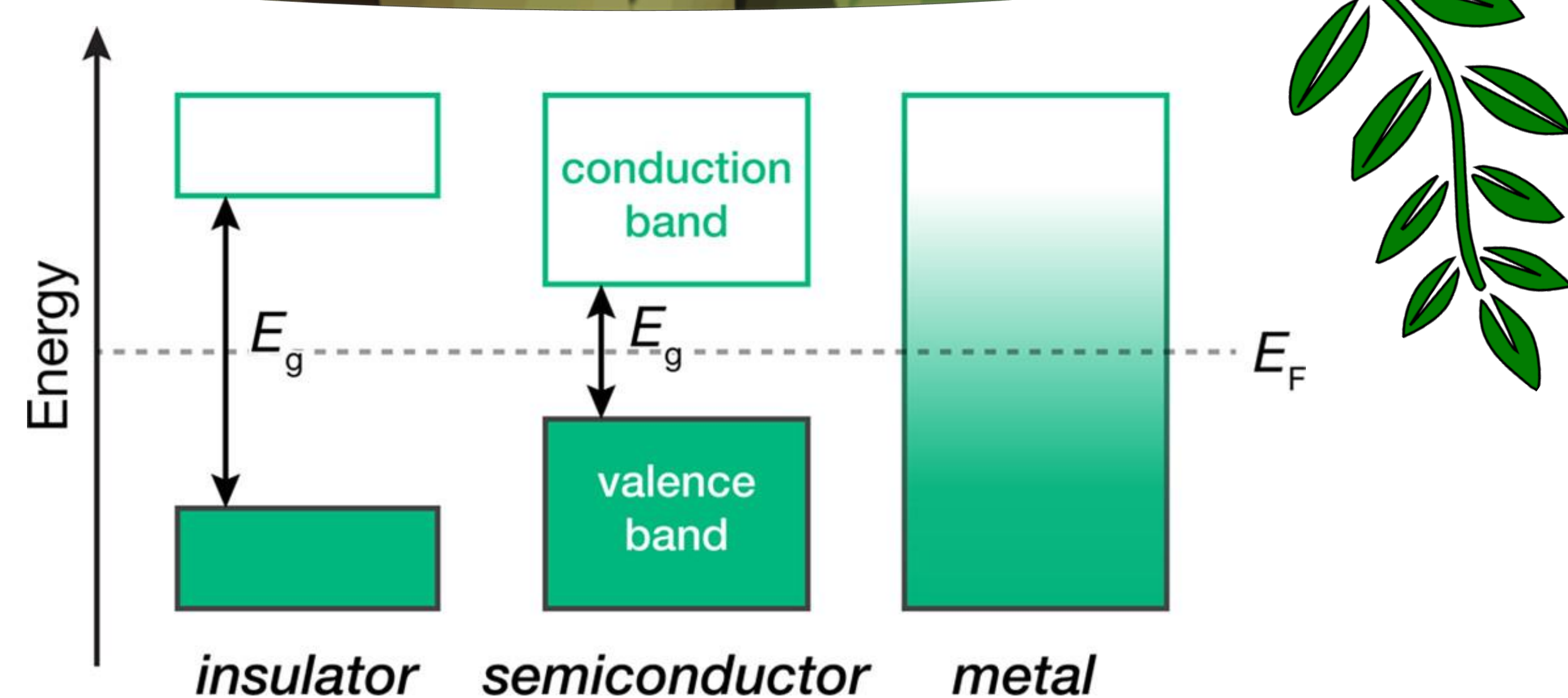
Ionic interaction prevent charge delocalization between the metal cluster and the organic linker → large band gap → Terrible conductor



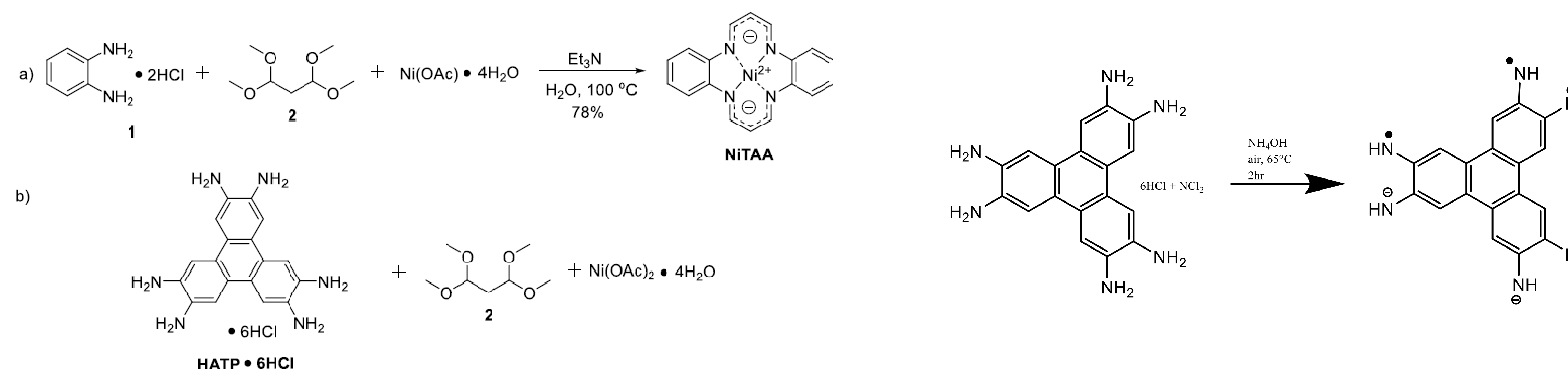
Conductivity: 40 S/cm

Conductivity: 10^{-10} S/cm

Band Structure



Linker formation



Results

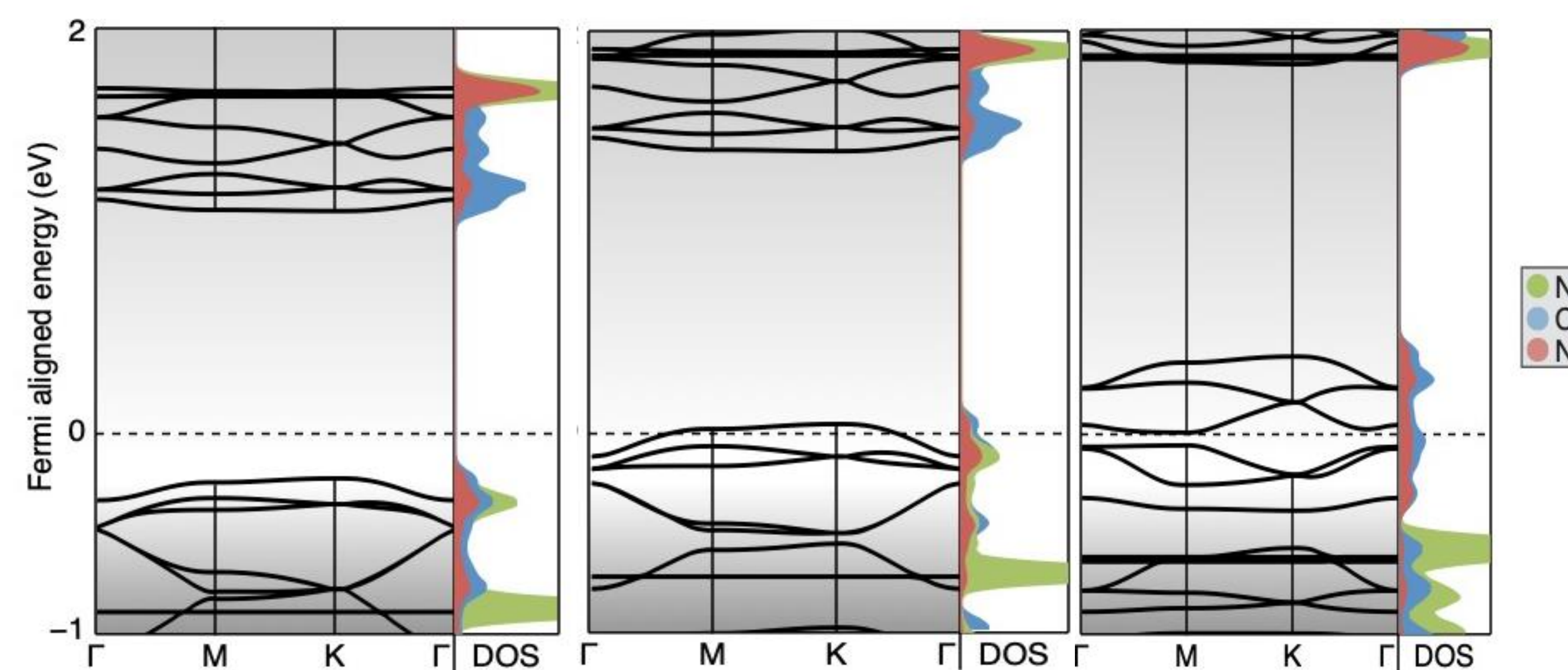


Figure 1. NiTAA-MOF monolayer band structure. (left) Neutral charge monolayer (middle) +1 charge monolayer (right) +6 charge monolayer

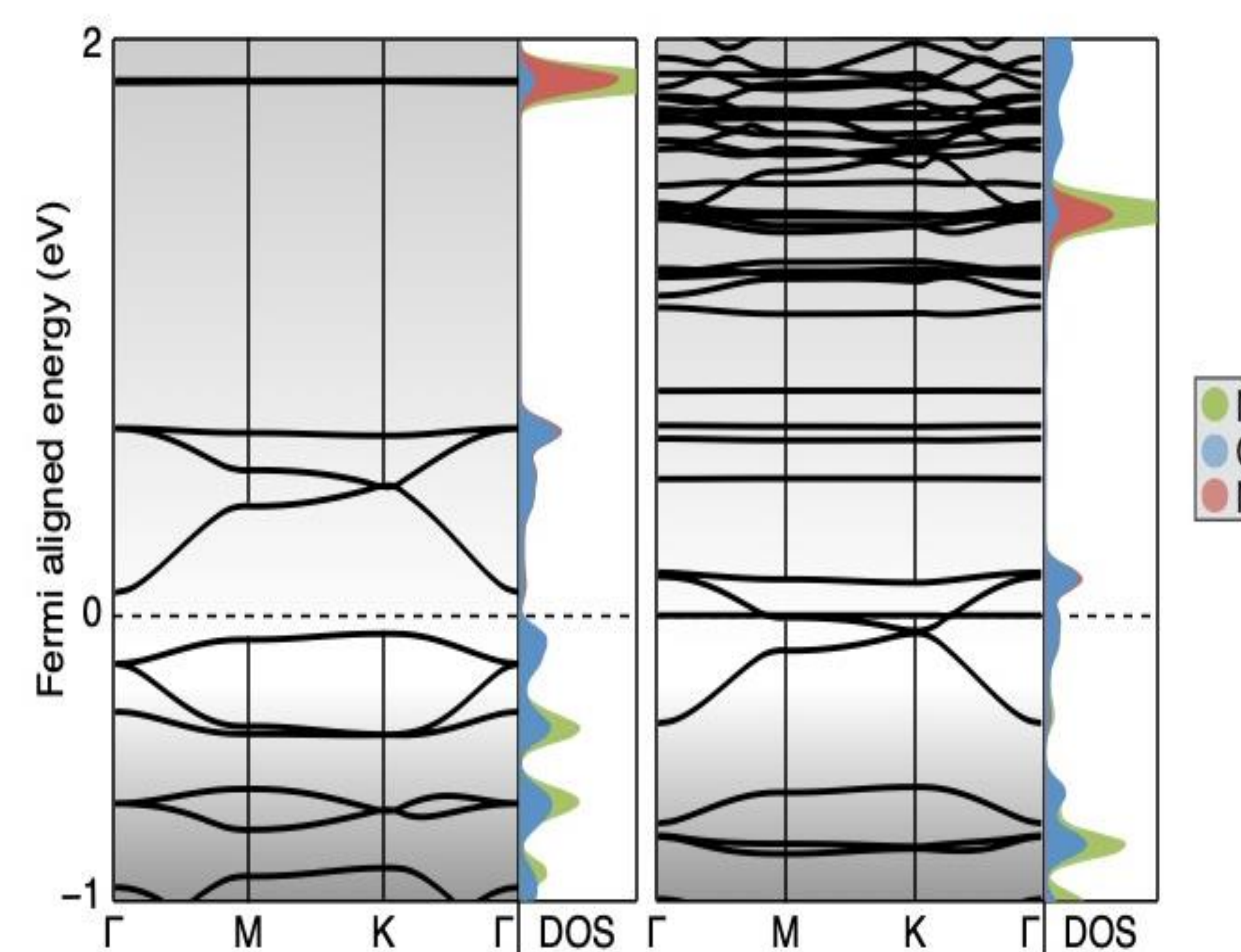


Figure 2. Ni₃(HITP)₂ monolayer band structure. (left) Neutral charge monolayer (right) -6 charge monolayer

Conclusion

Based on computational band structure result, Neutral charge NiTAA-MOF is an insulator, while +1 and +6 showing metallic feature and semi-conductor feature, respectively. While HITP linker truncate oxidation state both remain unoxidized between NiTAA-MOF⁺⁶ and Ni₃(HITP)₂⁻⁶, NiTAA-MOF⁺⁶ show semi-conductor behavior while Ni₃(HITP)₂⁻⁶ shows metallic behavior. Bulk material and effect of metal exclusion still need to be further studied.

Materials and methods

All calculations were performed within the Kohn-sham DFT framework as implemented in Vienna ab initio simulation package (VASP). Optimization calculation of NiTAA and Ni₃(HITP)₂ monolayer was both optimized under ionic relaxation of NSW=100 and ISIF=2. KPOINTS=auto while bulk material was done under ISIF=3. Electron density map was obtained through LPARD=TRUE based on optimized geometry of each model. Band structure calculation were done under PBEsol exchange-correlation functional with Grimmes dispersion (IVDM=12) factor included.

Reference

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J. Am. Chem. Soc. 2014, 136, 25, 8859–8862
Chem. Rev. 2020, 120, 16, 8536–8580

Acknowledgments

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